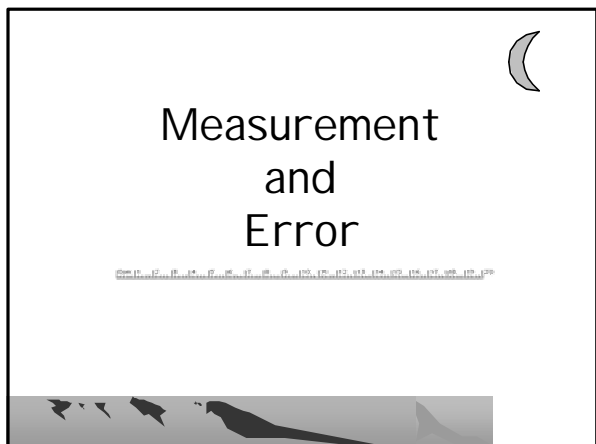
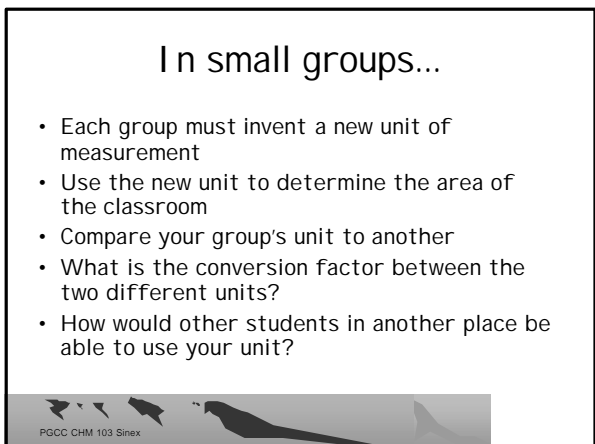


Measurement and Error




In small groups...

- Each group must invent a new unit of measurement
- Use the new unit to determine the area of the classroom
- Compare your group's unit to another
- What is the conversion factor between the two different units?
- How would other students in another place be able to use your unit?



Can you hit the bull's-eye?

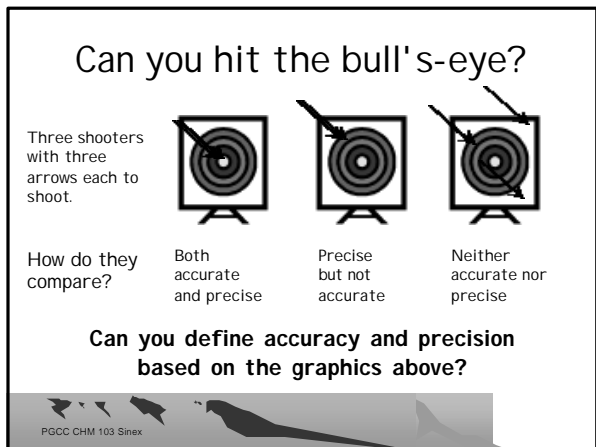
Three shooters with three arrows each to shoot.



How do they compare?

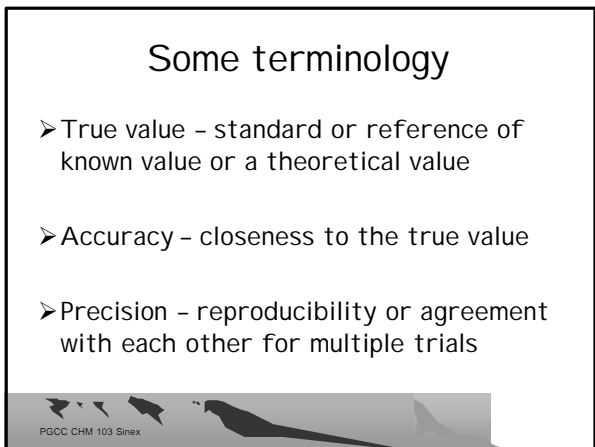
| | | |
|---------------------------|--------------------------|------------------------------|
| Both accurate and precise | Precise but not accurate | Neither accurate nor precise |
|---------------------------|--------------------------|------------------------------|

Can you define accuracy and precision based on the graphics above?

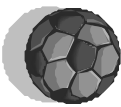


Some terminology

- True value - standard or reference of known value or a theoretical value
- Accuracy - closeness to the true value
- Precision - reproducibility or agreement with each other for multiple trials



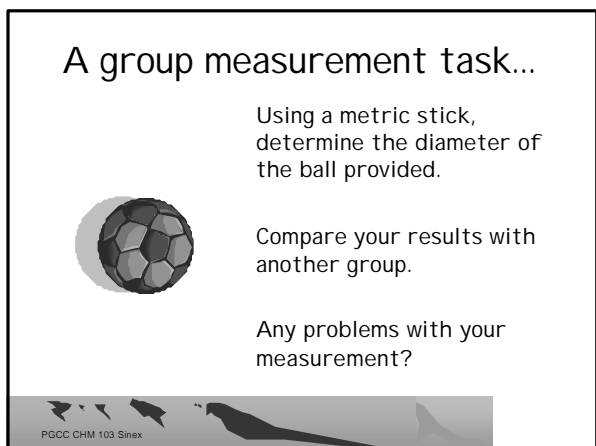
A group measurement task...



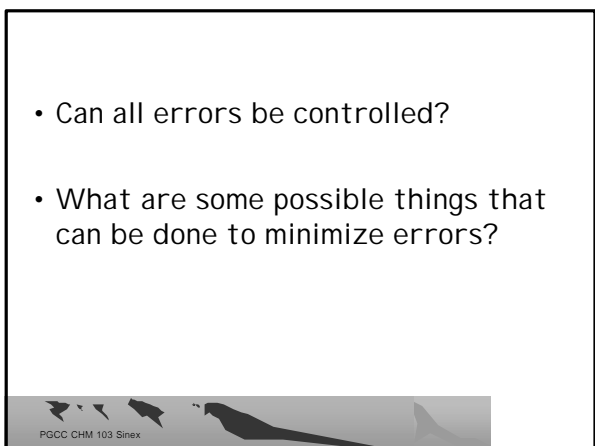
Using a metric stick, determine the diameter of the ball provided.

Compare your results with another group.

Any problems with your measurement?



- Can all errors be controlled?
- What are some possible things that can be done to minimize errors?



Types of Errors

- Determinate (or Systematic)
 - Sometimes called bias due to error in one direction- high or low
 - Known cause
 - Operator
 - Calibration of glassware, sensor, or instrument
 - When determined can be corrected
 - May be of a constant or proportional nature

constant or proportional error...

Proportional error influences the slope.

Constant error influences the intercept.

Types of Errors continued

- Indeterminate (or Random)
 - Cannot be determined (no control over)
 - Random nature causes both high and low values which will average out
 - Multiple trials help to minimize

Sources of Error

sampling

Representative sample homogeneous vs. heterogeneous

How about sampling a chocolate chip cookie?

→ preparation →

Loss

Contamination (unwanted addition)

analysis

Measurement of Analyte

Calibration of Instrument or Standard solutions

How do you assess the total error?

- One way to assess total error is to treat a reference standard as a sample.
- The reference standard would be carried through the entire process to see how close the results are to the reference value.

Expressing accuracy and precision

- Mean (average)
- Percent error


} accuracy

- Range
- Deviation
- Standard deviation
- Percent coefficient of variation

} precision

Accuracy and Precision

The center of the target is the true value.




| | | | |
|----------------------------------|---------------------------|--------------------------|-------------------------------------|
| Nature of accuracy and precision | Both accurate and precise | Precise only | Neither accurate nor precise |
| Target shooters comments | Great shooting! | Gun barrel must be bent! | Can't hit the broad side of a barn! |

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| | Both a & p | P only | Neither a nor p |
|---------------------|--|---|---|
| Scientific comments | <ul style="list-style-type: none"> • Very small error in measurement • All cluster the true value • Remember a standard or true value is needed | <ul style="list-style-type: none"> • Clustered multiple measurements but consistently off from true value • Calibration of probe or other measuring device is off or unknown systematic error | <ul style="list-style-type: none"> • The shot-gun effect • Get a new measurement system or operator |

PGCC CHM 103 Sinex

| | Both a & p | P only | Neither a nor p |
|-----------------------|---|---|---|
| Mathematical comments | <ul style="list-style-type: none"> • Small standard deviation or %CV • Small %error | <ul style="list-style-type: none"> • Small standard deviation or %CV • Large %error | <ul style="list-style-type: none"> • Large standard deviation or %CV • Large %error |
| Status/rank | Highly desired  | Acceptable, the best that can be done without a true value | Unacceptable |

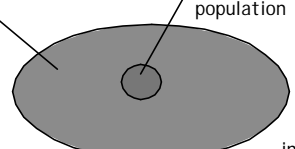
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- How would practice in target shooting or making a measurement influence the results?
- To judge the accuracy of an analysis, what is required?

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Standard Deviation of the...

- Population
- Sample - part of population



Estimates the variation in the population

Actual variation in the population

May not be representative sample

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A sampling activity

- From the bag of poker chips (THE POPULATION), containing two colors, pull out FIVE chips (THE SAMPLE). Record your results and calculate the %white.
- Place your five chips back into the bag.
- Repeat the five chip sample four times.

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When the chips are down!

- How would you characterize the results of the multiple sampling?
- Do you need to take more samples of five chips?
- Would collecting a larger sample, say 10 chips, improve your results?

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Population

$$\sigma_{\text{population}} = \sqrt{\frac{\sum(\text{deviations})^2}{n}}$$

Sample

$$\sigma_{\text{sample}} = \sqrt{\frac{\sum(\text{deviations})^2}{n - 1}}$$

- THE SAMPLE STANDARD DEVIATION allows for more variation in the sample compared to the population, since sample is only part of population. Dividing by n-1 increases the estimate of the population variation. This attempts to eliminate the possibility of bias.
- On the TI-83, the sample standard deviation is given as Sx when you do 1-Var Stats under [STAT] CALC.

In Excel, use STDEV under insert function, fx

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Graphical methods

- Scatter plots

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- Box and whisker plot on TI-83

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Let's examine some results for two students analyzing two different CaCO₃ antacid tablets

| True value | Student 1 | Student 2 |
|-------------|-----------|-----------|
| Label value | 500 mg | 750 mg |
| Mean | 463 mg | 761 mg |
| Std. dev. | 20 mg | 28 mg |

Which student has the more accurate results?
Which student has the greater precision?

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How are we going to address these questions?

| | | |
|----------|-----------|-----------|
| quantity | Student 1 | Student 2 |
|----------|-----------|-----------|

%CV

%Error

What does the sign of the percent error tell you about the error?

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and always remember to...

- Make all measurements carefully and check your results or readings a second time.
- Read all devices to as many places as possible (significant figures):
 - calibration marks + one more place
 - A buret, which is calibrated to 0.1 mL, can be read to 0.01 mL.
 - A thermometer marked every degree can be read to 0.1 degree

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