1. Determine the number of significant figures in each of the following measurements.
   a. 6006 mm  
   b. 0.056700 g  
   c. 7.800 x 10^3 min  
   d. 0.000064 kg  
   e. 7.09 x 10^4 mcg  
   f. 0.1040 mL

2. How many kg are in 3.56 x 10^4 mcg?

3. How many mg of silver are in 0.023 ounces of silver? 16 oz = 1 lb, 454 g = 1 lb

4. Decide if each of the following statements represents a fact, law, theory, hypothesis or belief. Explain your selection.
   a. The average velocity of any molecule is inversely related to its mass.
   b. The pressure of a gas increases with temperature because gaseous particles moving with a higher velocity make more frequent collisions resulting in greater force exertion.
   c. Chlorine is a highly reactive element.

5. If you travel at a posted speed limit of 65 mi/hr, how fast are you traveling in km/hr? in m/s?
   1.61 km = 1 mi

6. How many grams of dry air are in a room that is 21.0 ft x 17.5 ft x 8.0 ft.? The average density of dry air is 1.168g/L. 30.48 cm = 1 ft

7. In the manufacture of polyethylene, 3.2 lb of initiator are used for 5.0 tons of product. How many grams of initiator would be needed to produce 3.98 x 10^6 kg of polyethylene? 454 g = 1 lb, 2.20 lb = kg

8. The length of a test tube was determined by three students and their results are shown below. The actual length is 211 mm.

<table>
<thead>
<tr>
<th>Student 1</th>
<th>Student 2</th>
<th>Student 3</th>
</tr>
</thead>
<tbody>
<tr>
<td>210 mm</td>
<td>215 mm</td>
<td>201 mm</td>
</tr>
<tr>
<td>211 mm</td>
<td>230 mm</td>
<td>202 mm</td>
</tr>
<tr>
<td>210 mm</td>
<td>201 mm</td>
<td>202 mm</td>
</tr>
<tr>
<td>211 mm</td>
<td>245 mm</td>
<td>201 mm</td>
</tr>
<tr>
<td>209 mm</td>
<td>222 mm</td>
<td>201 mm</td>
</tr>
</tbody>
</table>

Which student(s) is precise? Which student(s) is accurate? Explain your selection.

9. The density of a sample of hydrochloric acid is 1.19 g/mL. What is the mass in grams of a 0.45 L sample of the acid?

10. An irregular sample of metal weighing 109.2 grams was placed in a graduated cylinder containing 21.0 mL of water. The volume registered in the cylinder after the metal was added was 33.2 mL. What is the density of the metal?

11. One baked potato provides an average of 31.0 mg of Vitamin C. If 5.0 lb of potatoes contain 15 potatoes, how many mg of Vitamin C are available per pound of potatoes?